

Synthetic Studies toward Sarain A. Formation of the Western **Macrocyclic Ring**

Moo Je Sung, Hyoung Ik Lee, Hee Bong Lee, and Jin Kun Cha^{*,†}

Department of Chemistry, University of Alabama, Tuscaloosa, Alabama 35487

jcha@chem.wayne.edu

Received December 27, 2002

Starting with the tricyclic core **2b**, annulation to form the 13-membered western ring of sarain A has been achieved to afford the macrocycle **30a** by initial construction of the sterically congested quaternary center at C-3, followed by elaboration of the C-3 side-chain and ring-closing olefin metathesis. Also included is a parallel conversion of tricycle 2c to macrocycle 30b containing a functionalized side-chain at N-1 suitable for attachment of the eastern macrocyclic ring.

Sarains A-C (1A-C) belong to a new family of structurally intricate alkaloids found in marine sponges. They were isolated by Cimino and co-workers from the sponge Reniera saraí^{1a,b} and reported to possess moderate antitumor, antibacterial, and insecticidal activity.^{1c} The biosynthesis of sarains has been postulated to involve reductive condensation of a bispyridinium macrocycle,² and their fascinating architecture has attracted considerable synthetic interest.³ The stereoselective synthesis of the tricyclic core of sarain A has been achieved by three groups and also in our laboratory.⁴⁻⁷ Although sarain A has not succumbed to a total synthesis to date, the Weinreb group reported the construction of the "western" macrocyclic ring by employing a Grubbs' ring-closing olefin metathesis strategy.^{4c} The "eastern" macrocyclic ring assembly of sarains was addressed by the Heathcock group in a synthesis of an appropriately designed model system.^{5b} Toward a total synthesis of **1A**, we herein report a successful annulation of a western macrocyclic ring onto





2b and 2c (Scheme 1) by adaptation of Weinreb's previous work.

Results and Discussion

By a slight modification of our previously reported synthesis of **2a**,⁷ we first secured multigram quantities of the N-Boc-protected tricyclic core **2b** in order to facilitate removal of the N-substituent at a later stage under mild conditions.⁸ Annulation to form the western macrocyclic ring onto 2b required initial construction of the quaternary center at C-3. Our initial plan was to utilize Weinreb's alkylation of a nitrile anion at C-3 (Scheme 2).^{4c} The nitrile **5** was prepared as an epimeric mixture via alcohol 4 by standard methods. Despite considerable experimentation, alkylation of 5 afforded none of the desired C-alkylation product 6; instead, the amide 7 was isolated in 50% (unoptimized) yield, presumably arising from N-alkylation. This disappointing result was in sharp contrast to Weinreb's previous observation that a nitrile bearing an exocyclic carbamate functionality (rather than an endocyclic lactam in 5) was amenable to C-3 alkylation; subtle, as yet unidentified,

[†]Current address: Department of Chemistry, Wayne State University, 5101 Cass Avenue, Detroit, MI 48202.

^{(1) (}a) Cimino, G.; Mattia, C. A.; Mazzarella, L.; Puliti, R.; Scognamiglio, G.; Spinella, A.; Trivellone, E. *Tetrahedron* **1989**, *45*, 3863. (b) Guo, Y.; Madaio, A.; Trivellone, E.; Scognamiglio, G.; Cimino, G. *Tetrahedron* **1996**, *52*, 8341. (c) Caprioli, V.; Cimino, G.; De Guilio, A.; Madaio, A.; Scognamiglio, G.; Trivellone, E. Comp. Biochem. Physiol. 1992, 103B, 293.

⁽²⁾ Cf. (a) Baldwin, J. E.; Whitehead, R. C. Tetrahedron Lett. 1992, *33*, 2059. (b) Baldwin, J. E.; Bischoff, L.; Claridge, T. D. W.; Heupel, F. A.; Spring, D. R.; Whitehead, R. C. *Tetrahedron* **1997**, *53*, 2271. (c) Gil, L.; Gateau-Olesker, A.; Marazano, C.; Das, B. C. Tetrahedron Lett. 1995, 36, 707. (d) Kaiser, A.; Billot, X.; Gateau-Olesker, A.; Marazano, C.; Das, B. C. J. Am. Chem. Soc. 1998, 120, 8026.

⁽³⁾ Matzanke, N.; Gregg, R. J.; Weinreb, S. M. Org. Prep. Proc. Int. 1998 30.1

^{(4) (}a) Sisko, J.; Weinreb, S. M. *J. Org. Chem.* **1991**, *56*, 3210. (b) Sisko, J.; Henry, J. R.; Weinreb, S. M. *J. Org. Chem.* **1993**, *58*, 4945. (c) Irie, O.; Samizu, K.; Henry, J. R.; Weinreb, S. M. *J. Org. Chem.* 1999, 64, 587.

^{(5) (}a) Henke, B. R.; Kouklis, A. J.; Heathcock, C. H. J. Org. Chem. 1992, 57, 7056. (b) Heathcock, C. H.; Clasby, M.; Griffith, D. A.; Henke, B. R.; Sharp, M. J. Synlett 1995, 467. (c) Denhart, D. J.; Griffith, D. A.; Heathcock, C. H. J. Org. Chem. 1998, 63, 9616.
 (6) Downham, R.; Ng, F. W.; Overman, L. E. J. Org. Chem. 1998, 62 (2007)

^{63. 8096.}

⁽⁷⁾ Sung, M. J.; Lee, H. I.; Chong, Y.; Cha, J. K. Org. Lett. 1999, 1, 2017.

⁽⁸⁾ Sung, M. J. Ph.D. Dissertation, The University of Alabama, December 2001.

SCHEME 2



electronic factors seem to play an important role in the successful nitrile alkylation.

The exclusive N-alkylation of the nitrile 5 prompted us to investigate N-allylation in tandem with a 3-aza-Claisen rearrangement of the presumed intermediate 8 (Scheme 3). Walters had shown that the Claisen rearrangement of 3-aza-1,2,5-hexatrienes could take place under exceptionally mild conditions.⁹ Unfortunately, in lieu of the 3-aza-Claisen rearrangement product 9, only amide 10 was obtained in 50% yield. When the amide 10 was subjected to Walters' rearrangement protocols [I₂/ (EtO)₃P/Et₃N] or other dehydrating reagents, it was recovered unchanged. Similarly, the Claisen rearrangement of allyl ether 12, which was readily prepared by palladium-mediated allylation of aldehyde **11**,¹⁰ was also futile under either thermal or Lewis-acid-catalyzed conditions. It became apparent that these approaches to the construction of the quaternary center at C-3 were thwarted by severe steric congestion.¹¹

Ultimately, treatment of aldehyde **11** (or its more stable epimer obtained upon silica gel column chroma-



tography) with an excess of aqueous formaldehyde in the presence of sodium carbonate gave diol 15a in 90% yield, along with minute amounts of the initial aldol product 14 (Scheme 4). Formation of 15a presumably involved a Tishchenko reaction and subsequent in situ hydrolysis of the resulting formate (structure not shown). The aldol condensation-Tishchenko reaction sequence could not be extended to other nonenolizable aldehydes such as 3-trimethylsilylpropargyl aldehyde and acrolein for easy installation of a longer side chain. Elaboration of the side chain of 15a proved to be challenging and required intramolecularity to overcome unfavorable steric effects. Toward this end, the two hydroxyl groups in 15a were first differentiated by the THP protecting group to provide an equal mixture of 16, 17, and 18. These were easily separated by column chromatography, and 18 was readily re-converted to 15a by a catalytic amount of p-TsOH in MeOH for recycling. Treatment of 16 with diethylphosphonoacetyl chloride (19) in the presence of pyridine, followed by removal of the THP protecting group with *p*-TsOH, afforded alcohol **20a**. Dess-Martin oxidation of 20a and subsequent intramolecular Wittig olefination of **21a** then provided δ -lactone **22a** in excel-

^{(9) (}a) Walters, M. A.; McDonough, C. S.; Brown, P. S., Jr.; Hoem, A. B. *Tetrahedron Lett.* **1991**, *32*, 179. (b) Walters, M. A.; Hoem, A. B.; Arcand, H. R.; Hegeman, A. D.; McDonough, C. S. *Tetrahedron Lett.* **1993**, *34*, 1453. (c) Walters, M. A.; Hoem, A. B. *J. Org. Chem.* **1994**, *59*, 2645.

⁽¹⁰⁾ Inoue, Y.; Toyofuku, M.; Taguchi, M.; Okada, S.-i.; Hashimoto, H. Bull. Chem. Soc. Jpn. **1986**, *59*, 885.

⁽¹¹⁾ Several other unsuccessful approaches are discussed in ref 8.

SCHEME 4







b: $R = (CH_2)_4OPMB$

To set the stage for building a 13-membered ring by ring-closing olefin metathesis, the saturated lactone **24a** was next prepared in 97% yield by hydrogenation of **22a** (Scheme 5). Following removal (91%) of the Boc protecting group by either TBSOTf or TFA, amide **25a** was obtained by acylation with 5-hexenoyl chloride in 75% yield. The other terminal olefin moiety was then introduced by selective reduction (75%) of the lactone with NaBH₄, followed by indium-mediated allylation¹³ (88%) of the resulting lactol **26a** to provide **27a** as a diastereomeric mixture. The key cyclization of the diene **27a** took place smoothly (a 0.4 mM solution in refluxing CH₂Cl₂) using Grubbs' second-generation catalyst¹⁴ to afford the 13-membered lactam **28a** in 71% yield. Dehydration of **28a** with the Martin sulfurane¹⁵ provided conjugated

(12) (a) Bestmann, H. J. Angew. Chem., Int. Ed. Engl. 1977, 16, 349.
(b) Bestmann, H. J.; Schobert, R. Tetrahedron Lett. 1987, 28, 6587.

CO₂Et

30a

⁽¹³⁾ Yi, X.-H.; Meng, Y.; Li, C.-J. *Tetrahedron Lett.* **1997**, *38*, 4731. (14) (a) Grubbs, R. H.; Miller, S. J.; Fu, G. C. *Acc. Chem. Res.* **1995**,

^{(14) (}a) Grubbs, R. H.; Miller, S. J.; Fu, G. C. Acc. Chem. Res. 1995, 28, 446. (b) Sanford, M. S.; Love, J. A.; Grubbs, R. H. J. Am. Chem. Soc. 2001, 123, 6543. (c) Trnka, T. M.; Grubbs, R. H. Acc. Chem. Res. 2001, 34, 18. For recent reviews, see: (d) Schuster, M.; Blechert, S. Angew. Chem., Int. Ed. 1997, 36, 2036. (e) Grubbs, R. H.; Chang, S. Tetrahedron 1998, 54, 4413. (f) Armstrong, S. K. J. Chem. Soc., Perkin Trans. 1 1998, 371.

^{(15) (}a) Arhart, R. J.; Martin, J. C. J. Am. Chem. Soc. 1972, 94, 5003. (b) Martin, J. C.; Franz, J. A.; Arhart, R. J. J. Am. Chem. Soc. 1974, 96, 4604.

SCHEME 6



diene **29a** in 60% yield at 88% conversion. For convenience, **27a**, **28a**, and **29a** were used as obtained in isomeric mixtures. These isomers complicated characterization but were inconsequential: hydrogenation (10% Pd/C) of **29a** gave the macrolactam **30a** (95%).

We next introduced an alkoxybutane N-substitutent in place of the PMB group prior to the construction of the pyrrolidinone ring so as to facilitate subsequent annulation of the eastern macrocyclic ring (Scheme 6). By this simple alteration in the amine starting material, large amounts of **15b** were prepared in comparable yields.⁸ The requisite differentiation of the two alcohols in 15b was more conveniently achieved by means of chloroacetylation compared to the above-mentioned use of the THP protecting group. The chloroacetate group was chosen over the acetate because of its facile removal by thiourea and Et₃N.¹⁶ Protection of **15b** with chloroacetic anhydride was selective for -OR1 (in preference to -OR²), but an excess of chloroacetic anhydride was used for rapid processing. Thus, sequential treatment of 15b with chloroacetic anhydride and diethylphosphonoacetyl chloride (19) gave an easily separable mixture of 31 (45%), 32 (30%), and 33 (5-10%). The last two compounds were recycled to afford **15b** by basic hydrolysis (K₂CO₃, MeOH) in nearly quantitative yield. Diene **27b** was then obtained by adaptation of the above-mentioned procedure for 27a; use of the Wilkinson catalyst was necessary for conversion of 22b to 24b to avoid concomitant hydrogenolysis of the PMB group. In contrast to uncomplicated ring closure of 27a, it was surprising that treatment of diene 27b with the Grubbs catalyst gave **28b** in lower (42%) yield, along with formation of a dimer. Silvlation of **27b** prior to ring-closing olefin metathesis gave a reliably higher yield of the desired macrocycle, and subsequent desilylation furnished **28b** in 54% overall yield. Upon treatment of 27b with TBSOTf, the secondary alcohol was selectively protected as the tert-butyldimethylsilyl ether, validating steric congestion around the primary alcohol. Finally, 28b was converted smoothly to the desired macrolactam **30b**, which we plan to utilize for installation of the eastern macrocyclic ring.

In summary, we have achieved the assembly of the western macrocyclic ring by initial construction of the sterically congested quaternary center at C3, followed by elaboration of the side-chain and ring-closing olefin metathesis. Refinement of the present approach to streamline the overall steps, as well as formation of the remaining eastern macrocycle in **30b**, is currently in progress to complete a total synthesis of sarain A.

Acknowledgment. We thank the National Institutes of Health (GM35956) for generous financial support.

Supporting Information Available: Experimental procedures and copies of ¹H and ¹³C NMR spectra of key intermediates. This material is available free of charge via the Internet at http://pubs.acs.org.

JO026914G

⁽¹⁶⁾ Use of triethylamine in place of NaHCO₃ was necessary: Naruto, M.; Ohno, K.; Naruse, N.; Takeuchi, H. *Tetrahedron Lett.* **1979**, *20*, 251.